

INTRODUCTION

Polyhydroxyalkanoates (PHA) are biological and biodegradable polyesters produced naturally through bacterial fermentation of lipids and sugars. Nowadays, they have attracted increasing interest as a sustainable alternative to conventional plastics for food packaging. However, there is limited information on their chemical composition.

OBJECTIVES

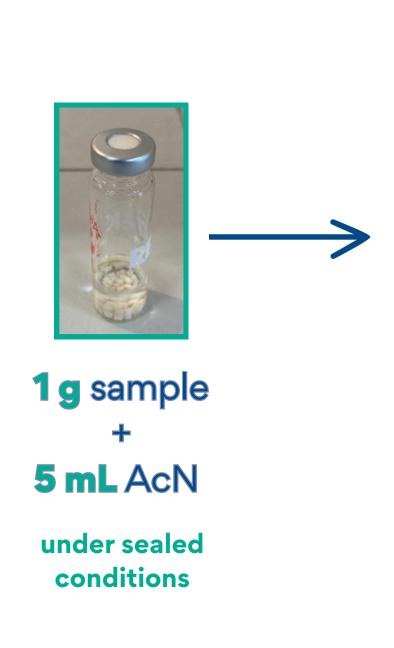
The aim of the present research work was :

- To characterise the non-volatile fraction of a PHA based material extracts using LC-MS.
- To estimate the toxicity of the tentatively identified oligomers by an "in-silico" method.



• Extraction procedure



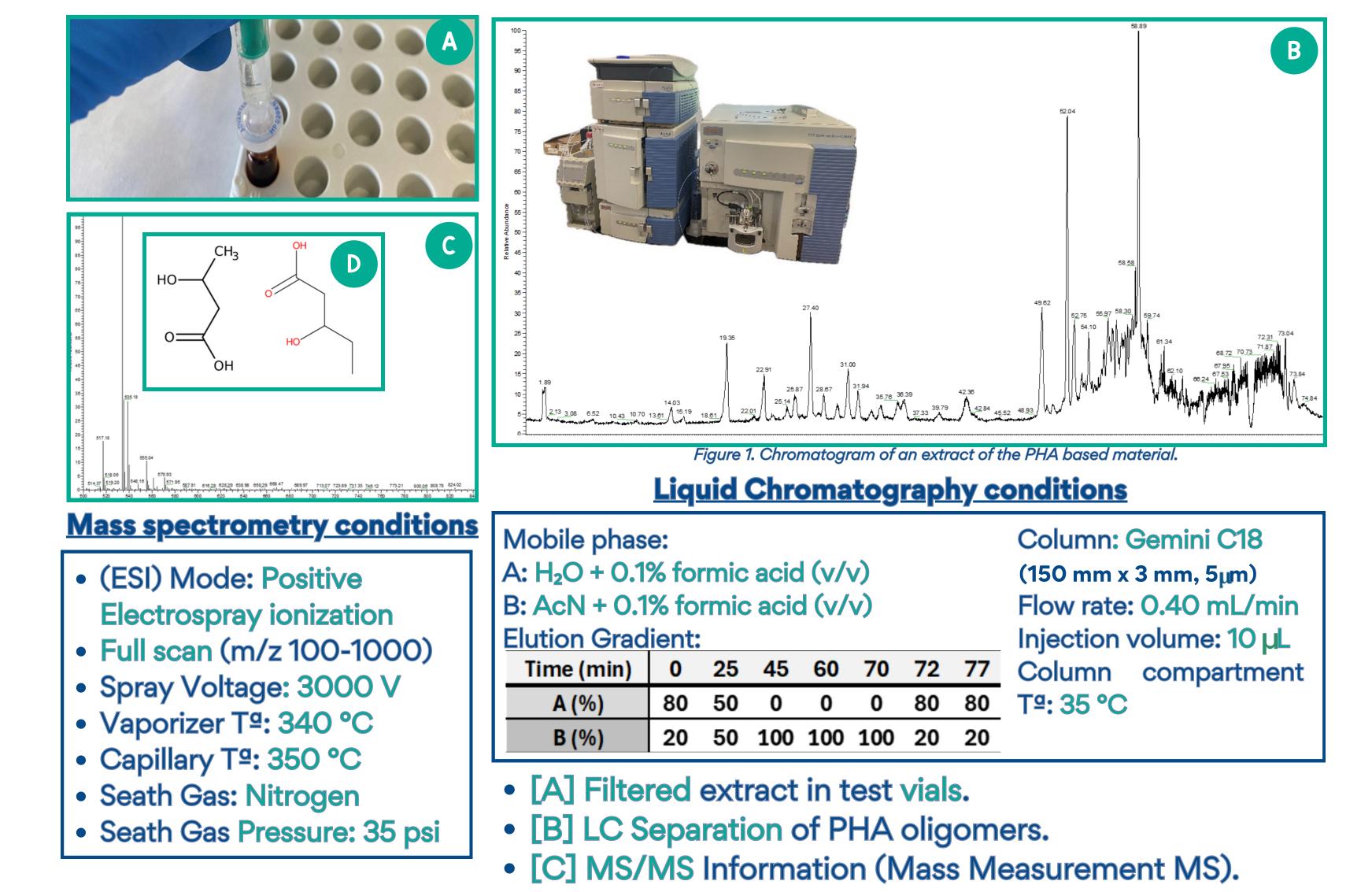


24 hours

at

70 °C

Toxicity estimation by ToxTree Software (based on Cramer's rules to classify into low, moderate and high toxicity (classes I, II and III respectively).



• [D] Tentative Chemical Identification (Suspicious

RESULTS

- The study was focused on low molecular weight oligomers (<1000Da), since it assumed that higher molecular compounds are not absorbed through the gastrointestinal tract.
- Oligomers of 3-hydroxybutyrate (3HB) ranging from the trimer (n=3) up to undecamer (n=11), and oligomers of 3-hydroxyvalerate (3HV) ranging from the trimer (n=3) up to nonamer (n=9) were identified [Table 1].
- Both series oligomers were tentatively identified in pellets and films.
- H, Na, K, NH_{A} and AcN main adducts were detected in both formats, indicating interactions of the oligomers with various agents (Choi et al., 2020; Hoque *et al.*, 2021).
- The theoretical toxicity of 3HB oligomers were Cramer I, while 3HV oligomers and 3HB oligomers (n=6 up to n=11) were Cramer III due to their

Table 1	Table 1 Table of identified oligomers. Oligomers detected in PHA samples (pellet and film).						
Tentative Oligomer type	Formula	Molecular Ions Observed (M/z)	Aducts	Toxicity (Cramer's rules)	Pellet Format	Film Format	
3 HB Trimer	C ₁₂ H ₂₀ O ₇	298.9, 315.0	[M+Na]+, [M+K]+	Ι	 Image: A start of the start of		
3HB Tetramer	$C_{16}H_{26}O_{9}$	363.1, 380.1, 385.0, 401.0	[M+H]+, [M+NH 4]+, [M+Na]+, [M+K]+	I	✓	✓	
3HB Pentamer	$C_{20}H_{32}O_{11}$	449.1, 466.1, 471.1, 487.1	[M+H]+, [M+NH 4]+, [M+Na]+, [M+K]+	Ι	✓	\checkmark	
3 HB Hexamer	$C_{24}H_{38}O_{13}$	535.1, 552.1, 557.1, 573.0	[M+H]+, [M+NH ₄]+, [M+Na]+, [M+K]+	III	 Image: A set of the set of the	✓	
3 HB Heptamer	$C_{28}H_{44}O_{15}$	638.8	[M+NH 4]+	III	 Image: A start of the start of	V	
3 HB Octamer	$C_{32}H_{50}O_{17}$	724.1, 729.0, 745.0	[M+NH 4]+, [M+Na]+, [M+K]+	III	 Image: A start of the start of	 Image: A set of the set of the	
→ 3 HB Nonamer	$C_{36}H_{56}O_{19}$	810.1, 831.3	[M+NH 4]+, [M+K]+	III	 Image: A start of the start of	✓	
3 HB Decamer	$C_{40}H_{62}O_{21}$	896.2, 901.05	[M+NH 4]+, [M+Na]+	III	 Image: A start of the start of		
3 HB Undecamer	$C_{44}H_{68}O_{23}$	982.2, 987.1	[M+NH 4]+, [M+Na]+	III	 Image: A start of the start of		
3 HV Trimer	$C_{15}H_{26}O_7$	319.1	[M+H]+	III	<		
3 HV Tetramer	$C_{20}H_{34}O_{9}$	436.2, 457.1	[M+NH ₄]+, [M+K]+	III		I	
3 HV Pentamer	$C_{25}H_{42}O_{11}$	557.2	[M+K]+	III		 Image: A start of the start of	
3 HV Hexamer	C ₃₀ H ₅₀ O ₁₃	657.5, 660.3	[M+K]+, [M+AcN]+	III		v	
3 HV Heptamer	C ₃₅ H ₅₈ O ₁₅	741.1, 757.2	[M+Na]+, [M+K]+	III		v	
3 HV Octamer	C ₄₀ H ₆₆ O ₁₇	860.3	[M+AcN]+	III		V	

greater structural complexity (Cramer *et al.,* 1978).

3 HV Nonamer C₄₅H₇₄O₁₉ 936.4

[M+NH ₄]+

~

III

CONCLUSIONS

- The PHA based material analyzed contains oligomers from 3 up to 11 units and from 3 up to 9 units of 3HB and 3HV, respectively.
- Adducts such as H, Na, K, NH₄ and AcN were detected, suggesting interaction with agents during the analysis of the PHA.
- Toxicity estimation shows 3HB oligomers (n=3 up to n=5) have low toxicity (Cramer I), 3HV oligomers and 3HB oligomers (n=6 up to n=11) exhibit high toxicity (Cramer III).
- Future work will be focused on migration testing.

REFERENCES

- Choi, YJ, Kim, DS and Kim, HJ (2020). Mass spectrometric analysis of oligomeric by-products in biodegradable polyhydroxyalkanoates (PHA). Journal of Analytical and Applied Pyrolysis, 149, 104841.
- Hoque et al., 2021 : Hoque, ME, Ghazali, S., & Park, JW (2021). Influence of chemical interactions on thermal and mechanical properties of polyhydroxyalkanoate oligomers. Polymer Degradation and Stability, 188, 109567.
- Cramer, F. H. (1978). Hydrocarbon oligomerisation: reactions and mechanisms. Journal of catalysis.

ACKNOWLEDGEMENTS

The study was financially supported by the Ministerio de Ciencia e Innovación, Agencia Estatal de Investigación and by Fondo Europeo de Desarrollo Regional (FEDER). Ref.Nº. PID2021-124729NB-IOO "MIGRABIOQUANT" (MCIN/AEI/ 10.13039/501100011033/FEDER, UE). Authors are grateful to "Ministerio de Ciencia, Innovación y Universidades" for the Predoctoral fellowship (ref. PRE2022-105027) awarded to CVG.